

Modeling Nickel-Silicidation using Physics-Informed Machine Learning

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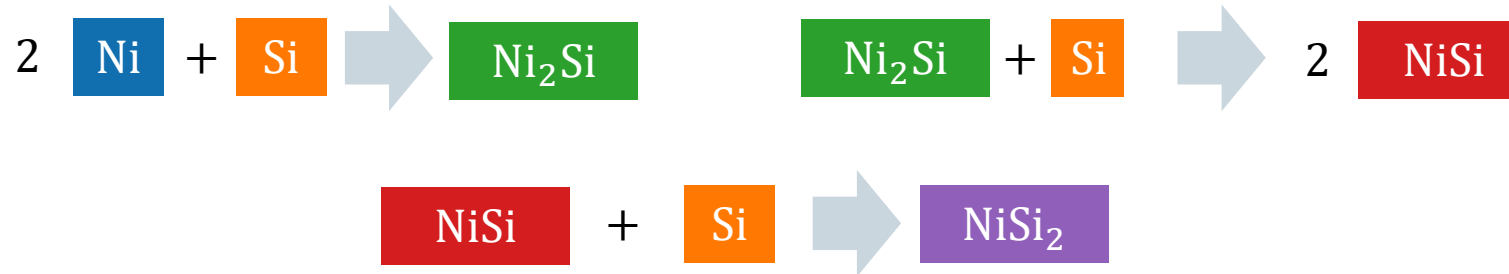
Problem motivation

- **Ohmic contact formation** fundamentally important for semiconductor technology (no contacts, no useful devices).
- Poorly understood, especially for novel materials such as silicon carbide.
- Contact quality highly **sensitive to**, e.g., **annealing temperature** and **duration**.
- Consider **nickel silicidation in silicon** as a **first step** towards proper silicidation models in silicon carbide.
- Analyse process by studying suitable nickel silicidation model and its solutions.

Nickel-Silicidation model

Chemical reactions and diffusion constants

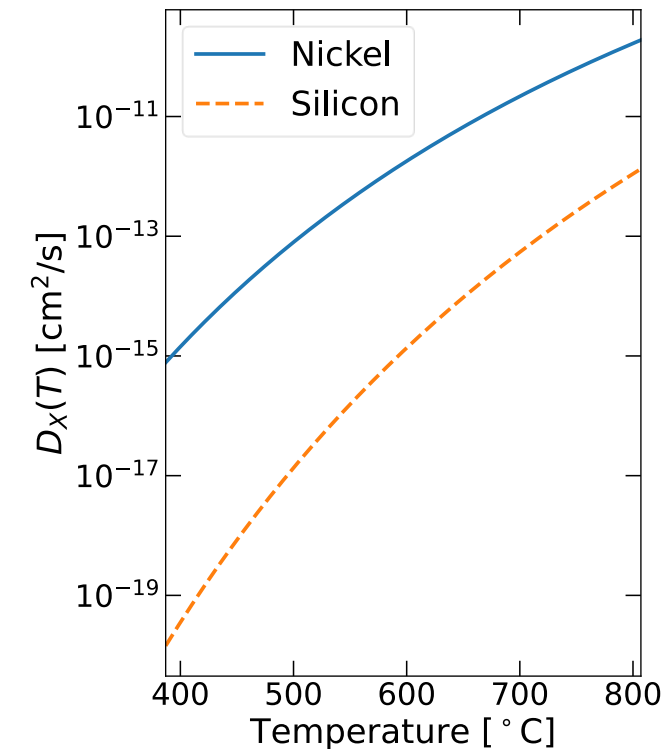
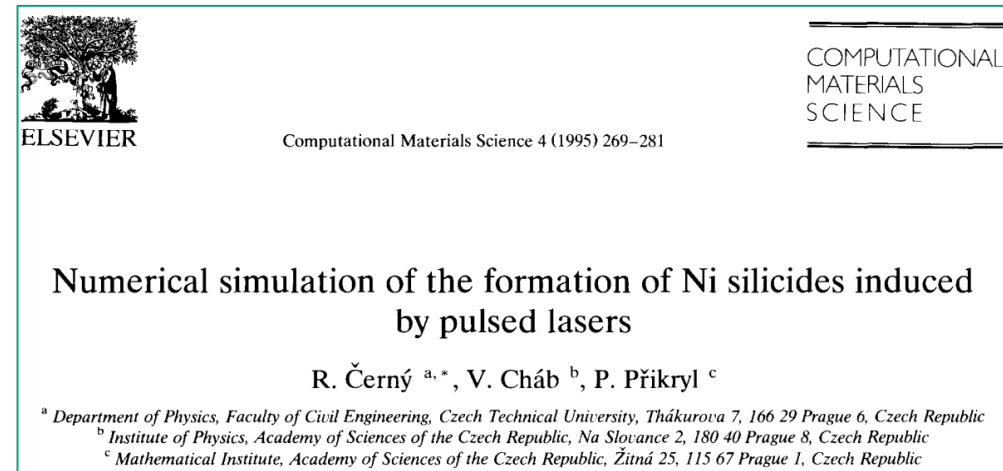
- Continuum model for Nickel-Silicidation adapted from Černý et al.
- Three implemented reactions:



- Diffusion of Ni and Si given by temperature dependent effective diffusion constants

$$D_X(T) = D_{X,0} \exp\left(-\frac{E_X}{k_B T}\right)$$

- Activation energy E_X and pre-factor $D_{X,0}$ were taken from literature sources.



Nickel-Silicidation model (2)

Reaction-diffusion system

- Concentration functions solve **reaction-diffusion PDE system**:

$$\frac{\partial C_{\text{Ni}}}{\partial t} = \nabla \cdot (D_{\text{Ni}} \nabla C_{\text{Ni}}) - 2k_1 \left(\frac{C_{\text{Ni}}}{C_{0,\text{Ni}}} \right)^2 C_{\text{Si}},$$

$$\frac{\partial C_{\text{Si}}}{\partial t} = \nabla \cdot (D_{\text{Si}} \nabla C_{\text{Si}}) - k_1 \left(\frac{C_{\text{Ni}}}{C_{0,\text{Ni}}} \right)^2 C_{\text{Si}} - k_2 \left(\frac{C_{\text{Ni}_2\text{Si}}}{C_{0,\text{Ni}_2\text{Si}}} \right) C_{\text{Si}} - k_3 \left(\frac{C_{\text{NiSi}}}{C_{0,\text{NiSi}}} \right) C_{\text{Si}},$$

$$\frac{\partial C_{\text{Ni}_2\text{Si}}}{\partial t} = k_1 \left(\frac{C_{\text{Ni}}}{C_{0,\text{Ni}}} \right)^2 C_{\text{Si}} - k_2 \left(\frac{C_{\text{Ni}_2\text{Si}}}{C_{0,\text{Ni}_2\text{Si}}} \right) C_{\text{Si}},$$

$$\frac{\partial C_{\text{NiSi}}}{\partial t} = 2k_2 \left(\frac{C_{\text{Ni}_2\text{Si}}}{C_{0,\text{Ni}_2\text{Si}}} \right) C_{\text{Si}} - k_3 \left(\frac{C_{\text{NiSi}}}{C_{0,\text{NiSi}}} \right) C_{\text{Si}},$$

$$\frac{\partial C_{\text{NiSi}_2}}{\partial t} = k_3 \left(\frac{C_{\text{NiSi}}}{C_{0,\text{NiSi}}} \right) C_{\text{Si}},$$

where $C_{0,X}$ denotes the equilibrium configuration of phase X .



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Numerical simulation of the formation of Ni silicides induced by pulsed lasers

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Nickel-Silicidation model (3)

Reaction rates and boundary conditions

- **Reaction rates**

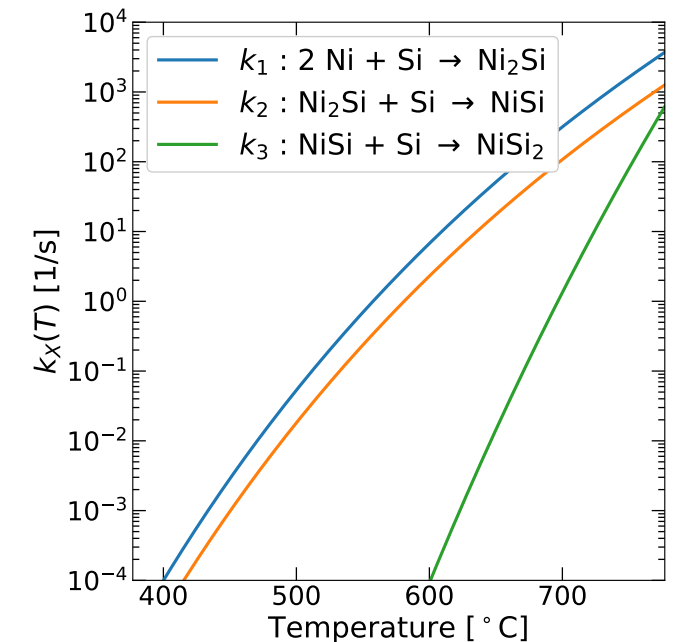
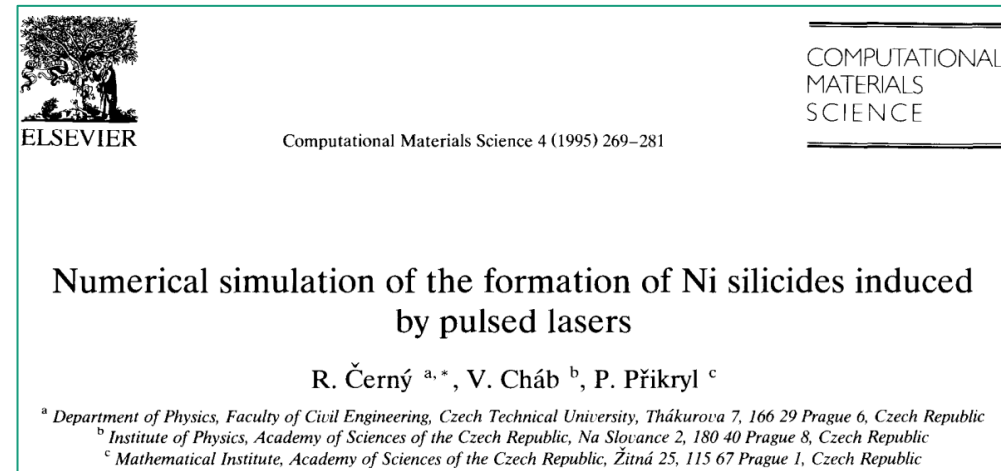
- We assume that reactions are diffusion-limited, such that

$$k_i \propto 4\pi r_i D_{\text{Si}}(T) \kappa(T; T_{i,c}).$$

- Here, r_i is the capture radius of the chemical reaction ($\sim 1 - 10 \text{ \AA}$).
- Each reaction happens only above a critical temperature $T_{i,c}$.
 - This is encoded in the cutoff function $\kappa(T; T_{i,c})$.

- **Boundary and initial conditions**

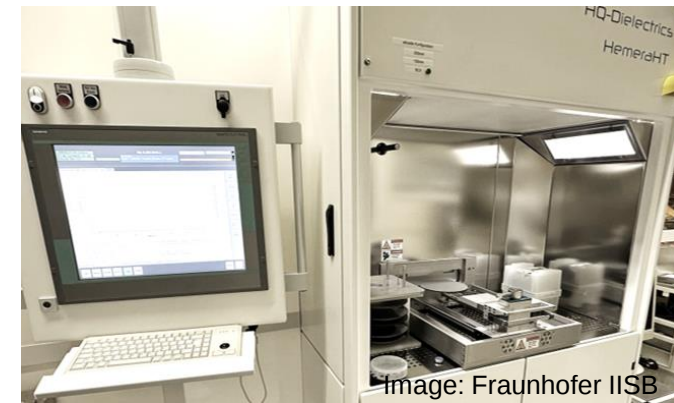
- Reflecting boundaries $\rightarrow$ modeled via vanishing Neumann boundary conditions.
- Initial conditions: $C_{\text{Ni}}(x, 0) = C_{0,\text{Ni}}$ within the Ni layer and $C_{\text{Si}}(x, 0) = C_{0,\text{Si}}$ within the Si domain.



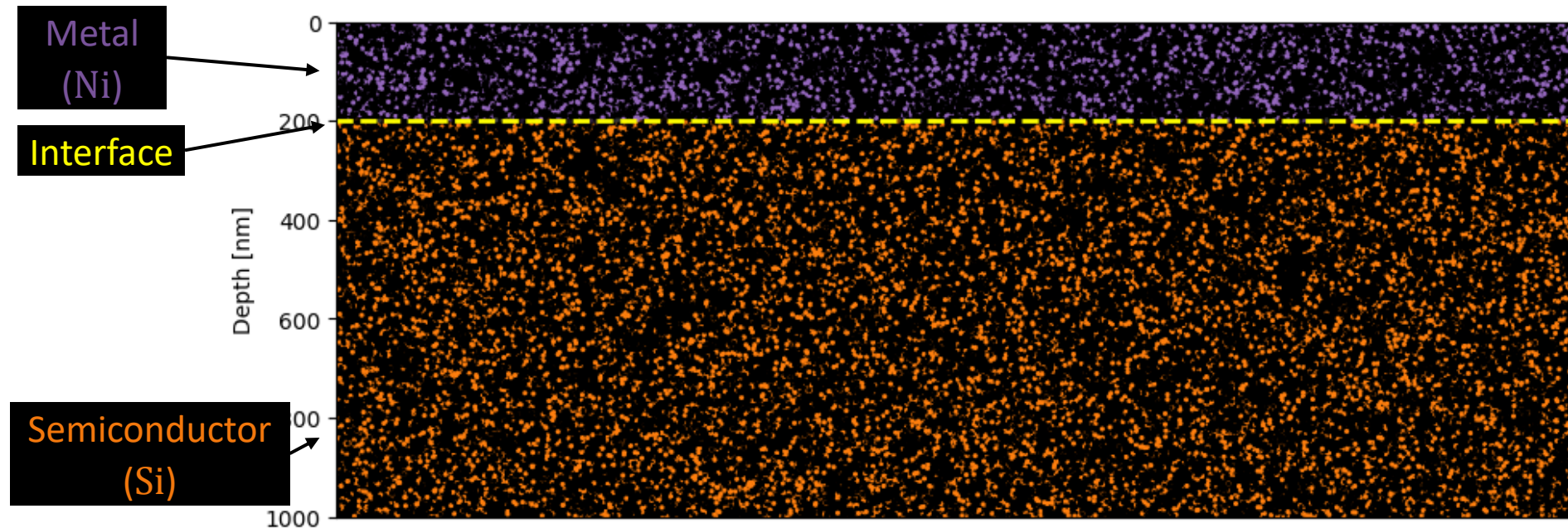
Flat setting Initial condition

Consider the following setting:

- Flat initial Ni-Si interface. → Model silicidation far away from edges of deposited Ni layer.
- Thickness of Ni layer: 200 nm. Depth of Si domain: 800 nm.



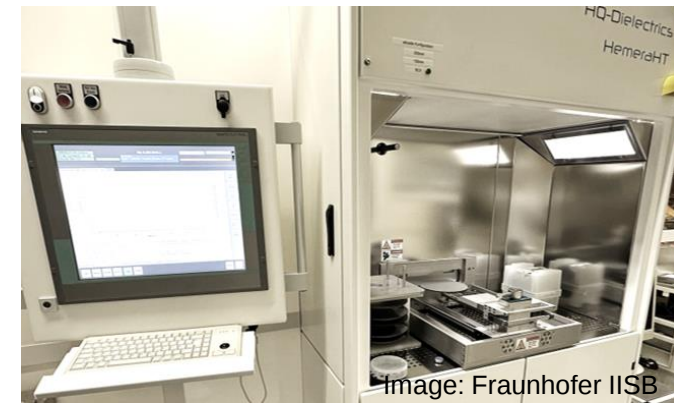
→ **Simulate RTP**



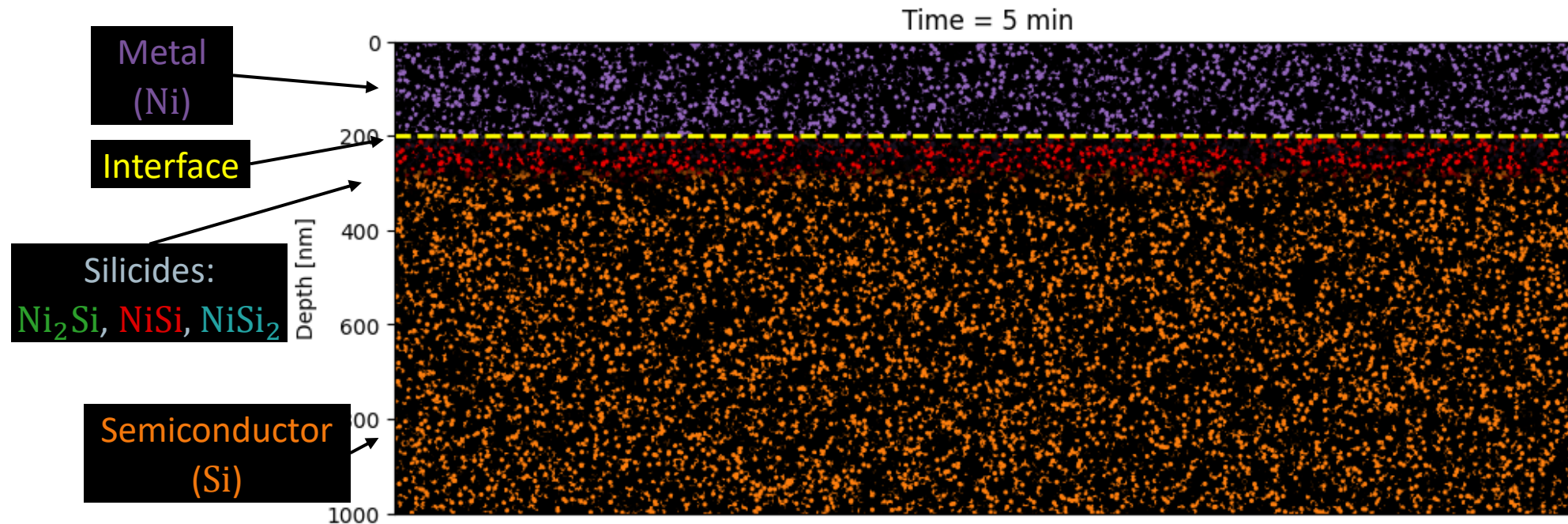
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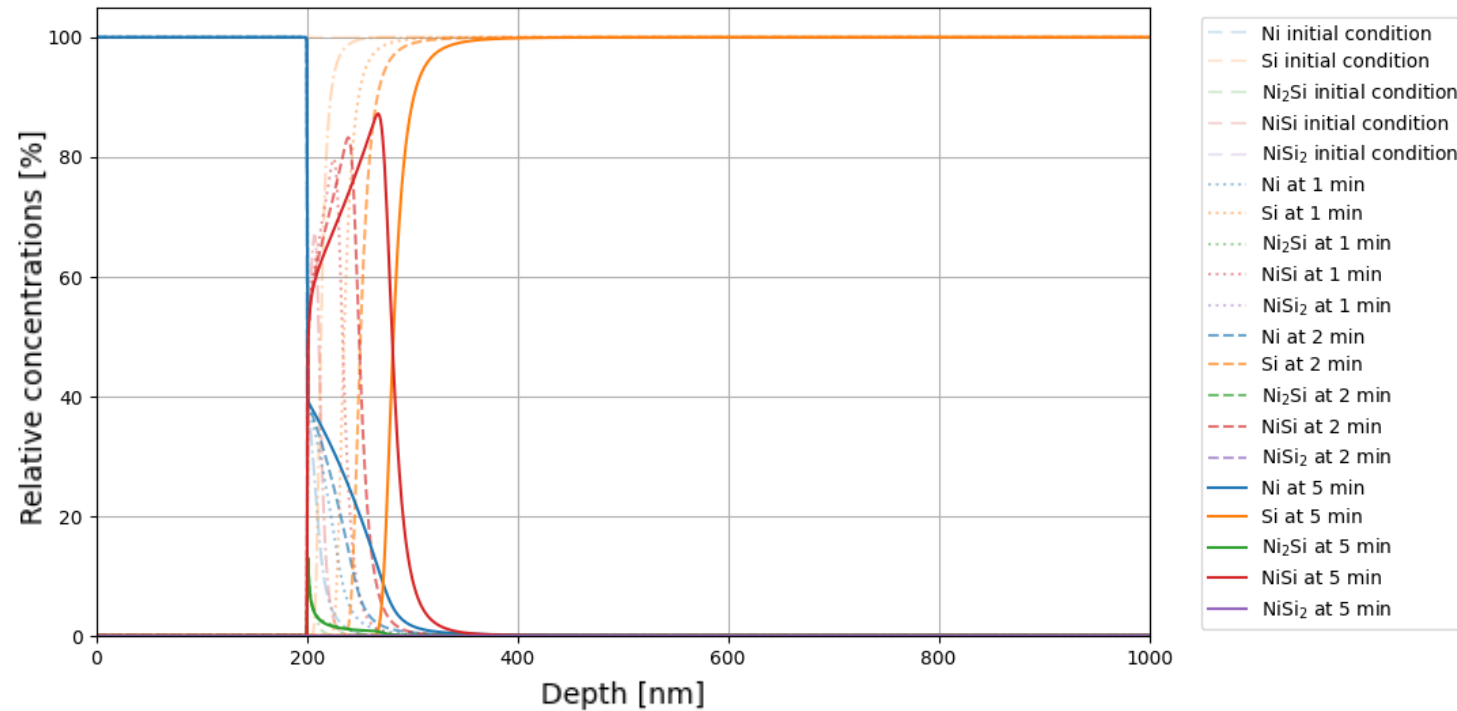
→ **Simulate RTP**



Flat setting

Evolution

- Consider the system for a maximum annealing time of 5 min.
- Model this setting via 1D concentration functions.



Flat setting

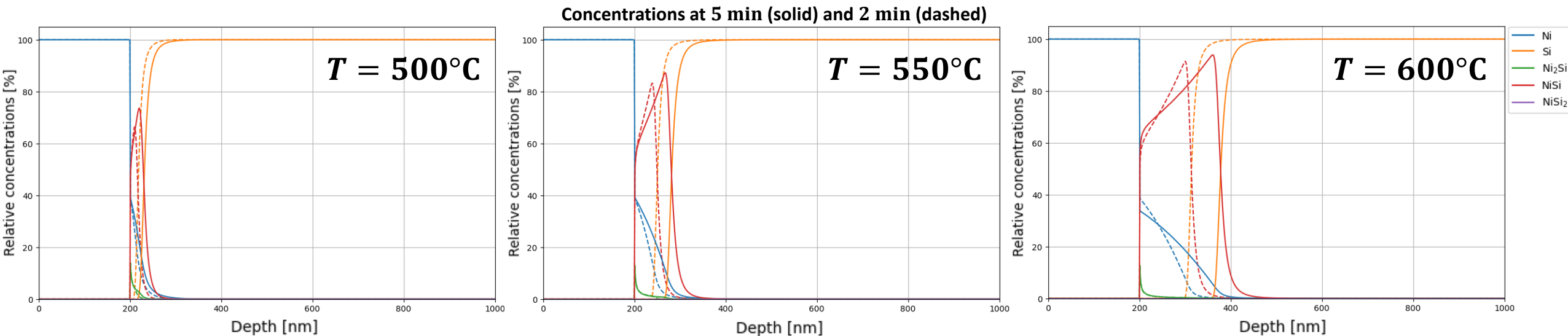
Final concentrations

With Synopsys' Sentaurus (commercial FEM solver):

A single computation takes

- ~ 25 s for 1D simulation,
- ~ 4 min 30 s for 2D simulation.

- Main quantity of interest: Final concentrations.
- So far: Considered temperature $T = 550^\circ\text{C}$. Different temperatures lead to different final concentrations.

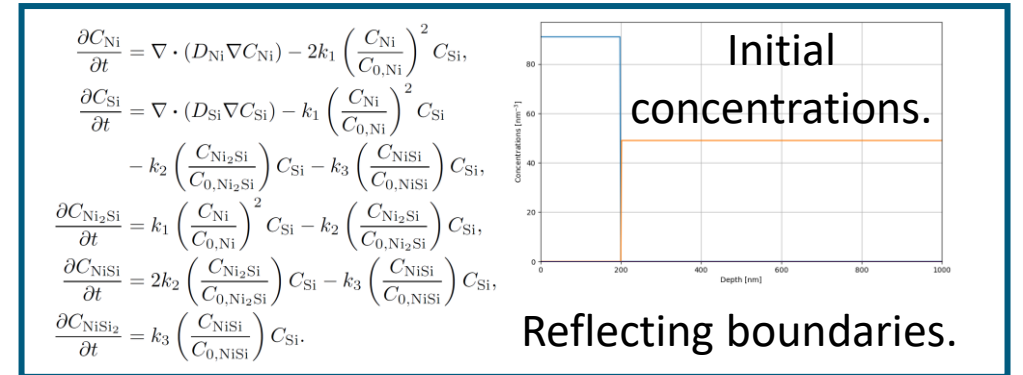
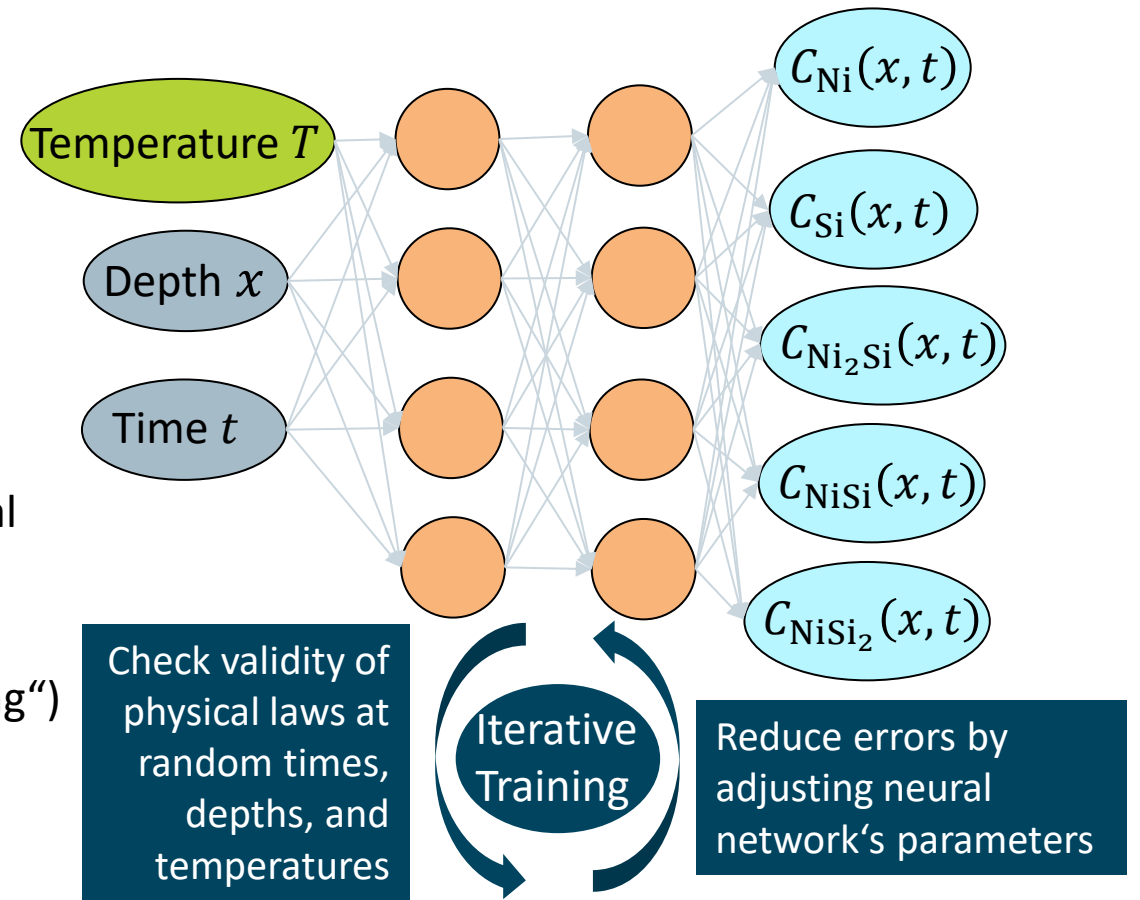


- Application of physics-informed machine learning: Model that can directly compute the concentrations for general temperatures and annealing times, without needing to rerun simulations for new settings.

Physics-informed machine learning

Main ideas:

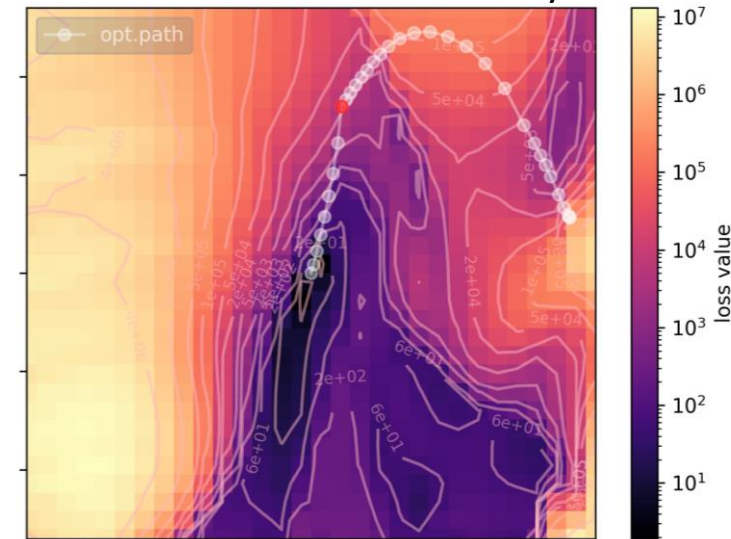
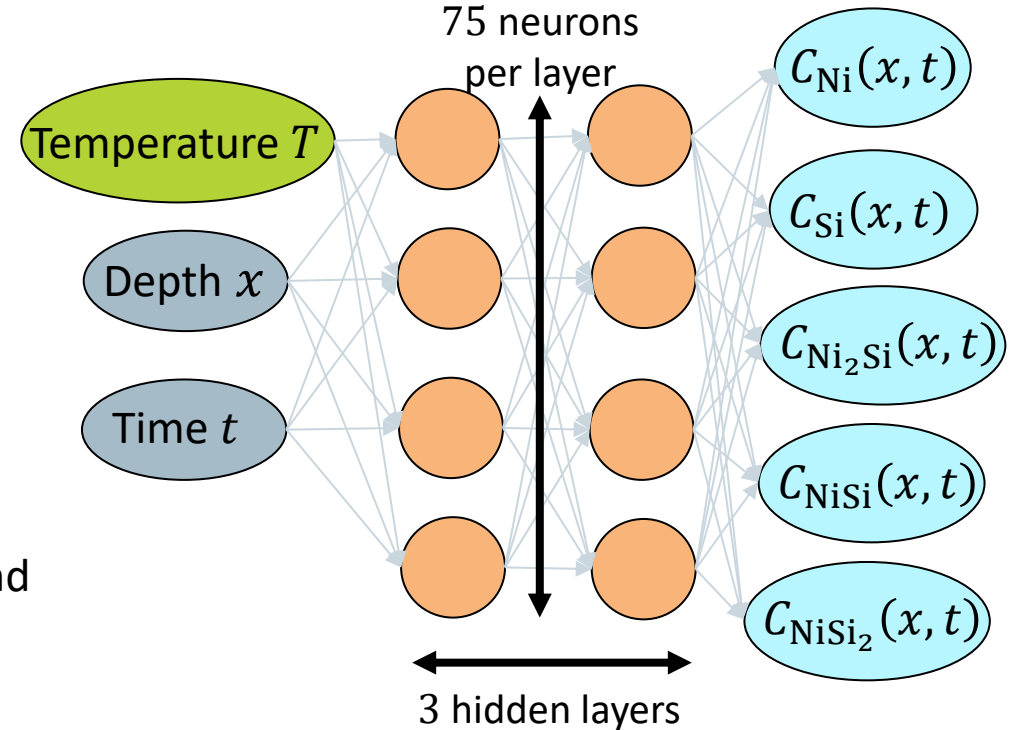
- Represent the solution of the silicidation process as a neural network.
- Optimize the neural network's internal parameters („training“) by minimizing the errors of the governing physical laws (PDEs, initial conditions, boundary conditions).
- No data needed for neural network training.
- Resulting model is a „**physics-informed neural operator**“ (PINO).



Physics-informed machine learning

Details

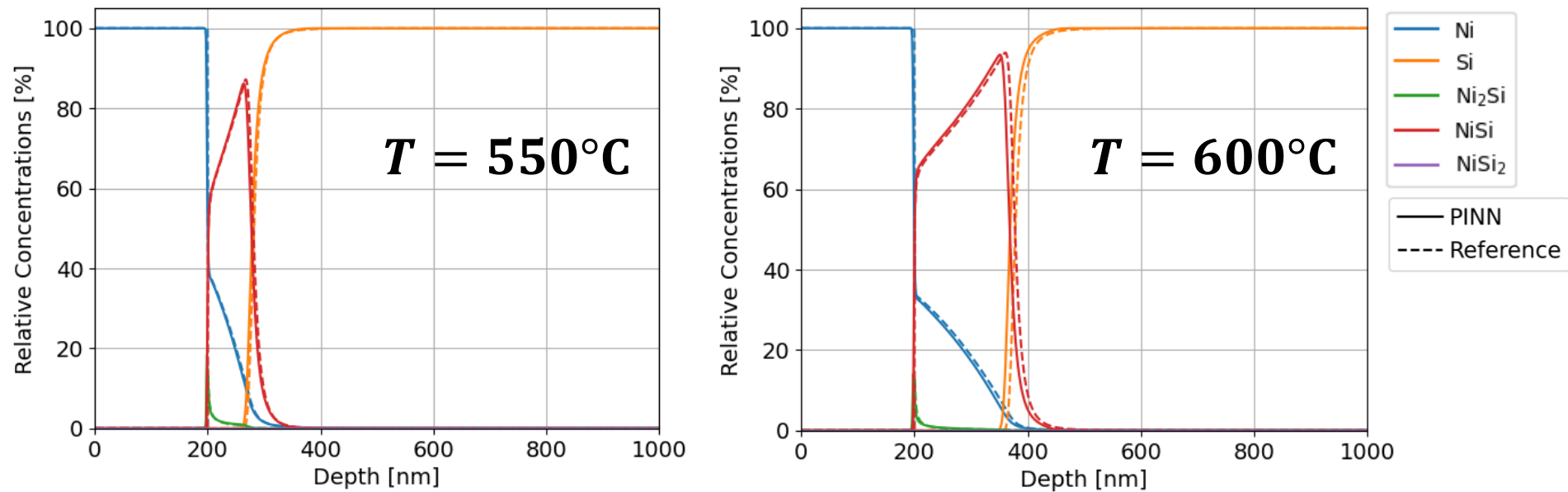
- **Neural network architecture:** MLP (multilayer perceptron / fully connected neural network) of size 3×75 .
- **Training:** $\sim 60\,000$ iterations with combination of Adam and L-BFGS optimizers.
→ Training time: ~ 3 h on a GPU (NVIDIA A100).
- **Collocation points:** Semi-adaptive sampling during training based on residuals („RAS“).
- **Hard-constraining:** Several properties directly included into neural network structure: all outputs ≥ 0 , upper bounds on concentrations, vanishing Neumann boundary conditions.
- **Implementation:** [DeepXDE](#) with backend  PyTorch



Results

Accuracy of PINO

Comparison of physics-informed neural operator with reference simulation at annealing time $t = 5$ min.



Similar accuracies for other temperatures:

Temperature	Error*
$T = 450^\circ\text{C}$	2.54 %
$T = 500^\circ\text{C}$	3.98 %
$T = 550^\circ\text{C}$	3.96 %
$T = 600^\circ\text{C}$	6.87 %
$T = 650^\circ\text{C}$	8.76 %

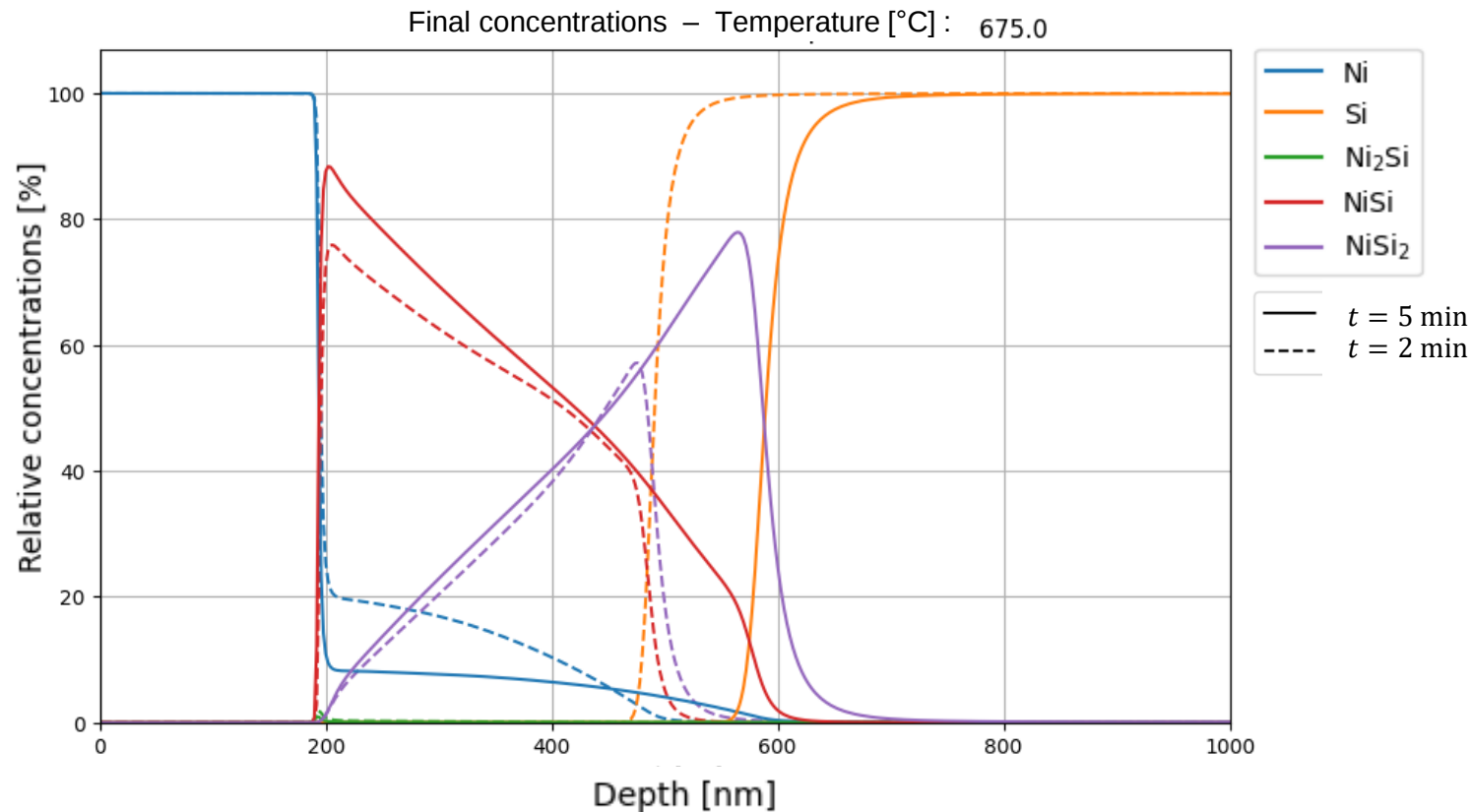
* Relative L^2 -distance to reference solution, averaged over components Ni, Si, and NiSi.

➤ PINO can accurately predict the silicidation of Ni in Si.

Results

General temperatures

Trained PINO can be evaluated for general temperatures and annealing times without retraining.



Computational costs to create animation:

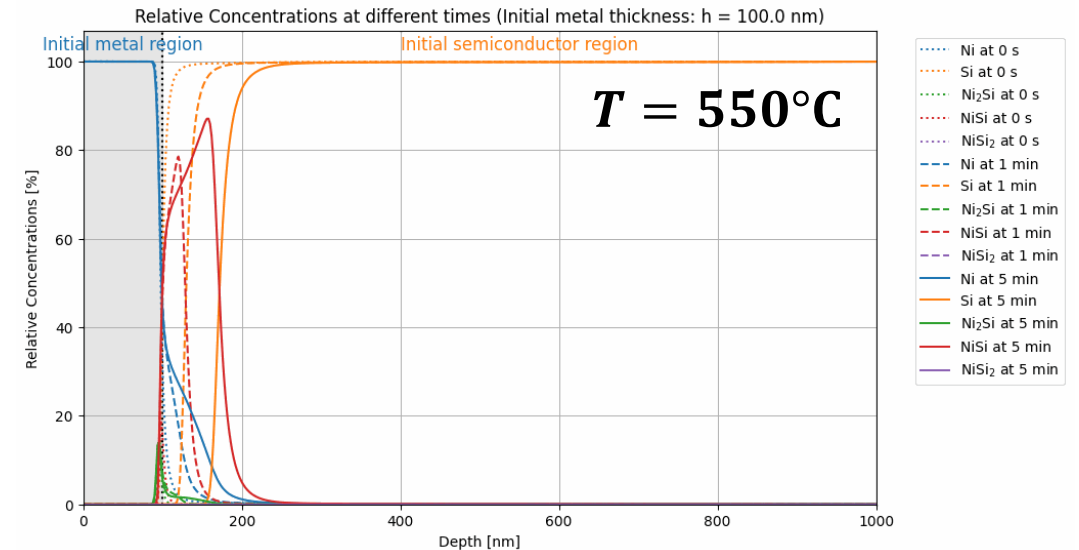
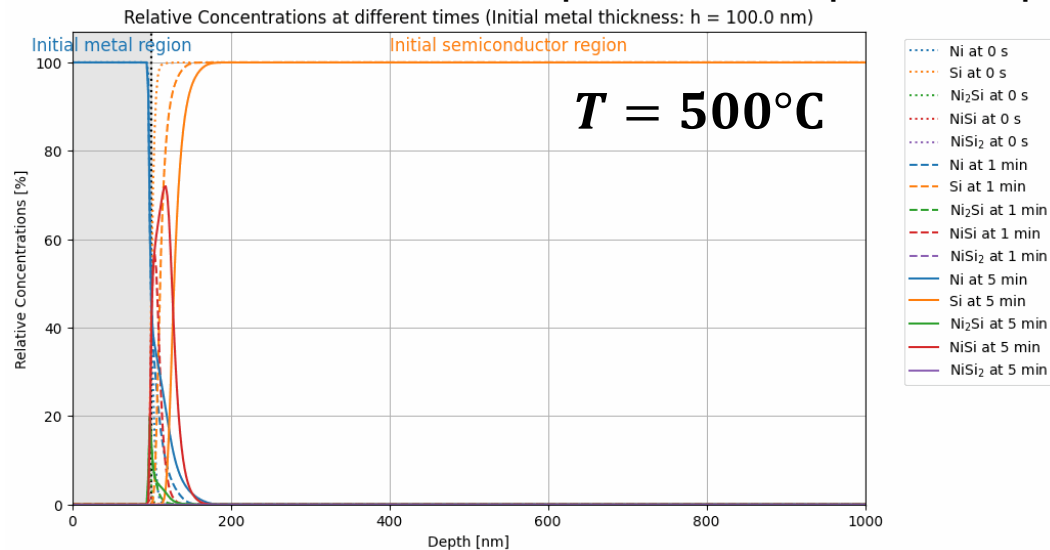
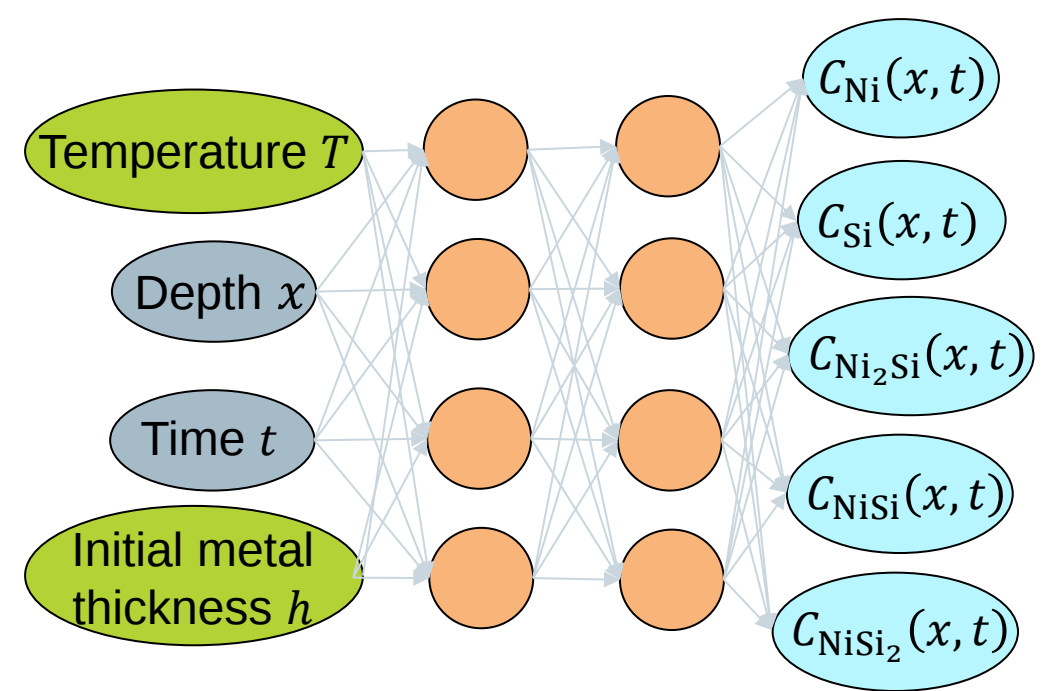
- Evaluation at 501 depth values (step size 2 nm) for 450 temperatures takes ~ **53 ms** on a laptop.
- Sequential classical simulations for 450 temperatures take ~ **3 h 5 min**.
→ Similar to model's training time.

Outlook

Further parameter dependencies

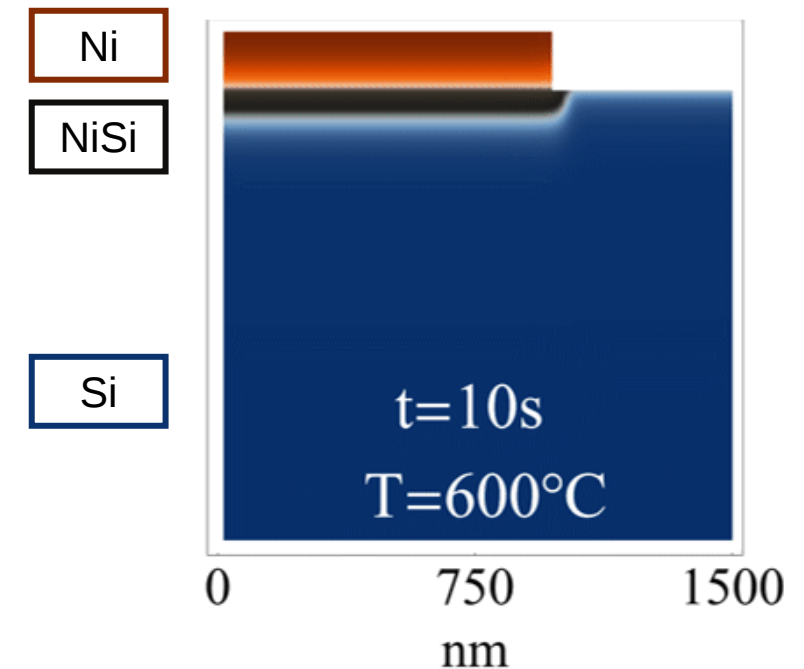
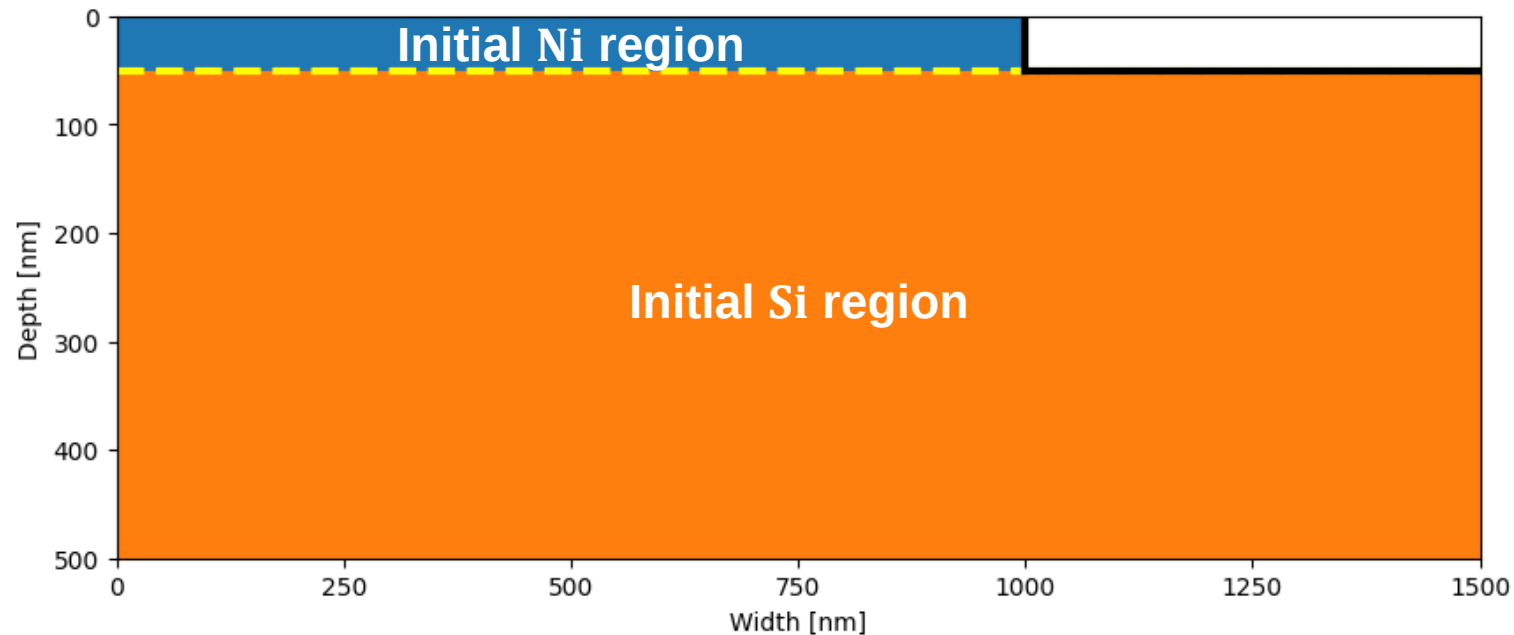
To offer fast optimization in parameters beyond temperature:
Extend parametric dependency of PINO.

- Also generalize over different initial geometries.
- In future: also allow time-dependent temperature profiles.



Outlook

- PINOs can be integrated in process control on edge devices.
- Extend PINO to 2D and 3D settings.



- Develop and calibrate model for Ni silicidation in SiC and train PINOs for it.

Contact

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